

*State University of Makassar*

**INTERNATIONAL CONFERENCE ON MATHEMATICS,  
SCIENCE, TECHNOLOGY, EDUCATION  
AND THEIR APPLICATIONS**

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Mathematics, Science, Technology, Education  
and their Applications"*

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APPLICATIONS

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## SCREENING OF TEMBELEKANG PLANT (*LANTANACAMARALINN*) ACTIVE COMPOUNDS FOR PREVENTION OF INFECTIOUS DISEASES IN SKIN WOUNDS

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### Abstract

*Lantana camara* Linn Plant is a plant that was developed as a potential source of secondary metabolites, which is, have potential as a drug based on ethnobotany approach. These plants are used especially as wound cure in the South Sulawesi. It was believed to cure very quickly in various types of skin wounds. In the previous study, the results of Chloroform extract of the *L. camara* Linn leaf showed high activity against *S. aureus* and *E. Coli* bacteria which each inhibition zone of 17.2 mm and 31.7 mm. But time to heal of *L. camara* Linn extract slower than patented drugs are often used. Therefore, the extract need to screening to eliminate inactive compounds. The extract was screened through several methods such as fractionation, purification and identification. Fractionation was done by flash column chromatography, purification by recrystallization and identification using color test, TLC and spectroscopic tests. The results showed there are three crystals that have been screened in the B fraction. These compounds react positively with Liberman-Buchard and  $\text{FeCl}_3$  reagent and showed orange color solution and a brown precipitate that identify as terpenoids and flavonoid group. This is supported by the data from IR spectrum of the compound.

**Keywords:** Lantana camara Linn leave; Plaster; Screening; Skin wounds.

### 1. Introduction

Indonesian society has made a series of efforts to prevent the disease using natural ingredients from a traditional medicine. Many people assume that the use of traditional medicine is relatively safer than synthetic or manufactured drugs. However, it does not mean traditional medicine does not have adverse side effects. This requires a more in-depth study on the chemical ingredients of natural materials and evidence of clinical efficacy for the use of traditional medicine is not only based on experience alone, but can be accounted for efficacy and safety are supported by scientific data.

Various types of plants with various medicinal purposes have been used as an alternative treatment for people who later supported by research from various

agencies. Some plants that have been popular as are ginger as a cure hepatitis and enteritis, java tea as a diuretic, turmeric as an antiseptic, arthritis and hepatitis.

Tembele kang plant (*L. camara* Linn) is a potent plant and very strategic was developed as a formula resource and tracking of secondary metabolites that have the potential as a drug. These herbs can be easily found in all regions, especially South Sulawesi. This plant in South Sulawesi is used as a cure wounds and is believed to cure various types of skin wound infection very quickly. Thus, this plant has allegedly metabolism of secondary metabolites as a potential antibacterial, so it can be developed for antibiotics and accelerate healing of wound infection.



acetate, methanol, chloroform, and other organic solvents such as chloroform and acetone. Several reagents such as Liebermann-Burchard reagent, Wagner, Dragendorff, FeCl<sub>3</sub> 1%, Whatman filter paper, aluminum foil, CeSO<sub>4</sub> 2%, tissue, silica gel, and aluminum-coated TLC plates of silica gel G 60 F<sub>254</sub>.

Research methods will include sample preparation, isolation of secondary metabolites of *L.camara* Linn, the best eluent determination by TLC, fractionation of extract by the best eluent use of column chromatography, monitoring analysis of fractions with the highest number of spots and separated by TLC to obtain fractions with the least number of spots, and purified by column chromatography and TLC to obtain pure compound. Furthermore, the identification of pure compounds with FTIR.

Sampling was carried out based on the location of plants *L. Camara* almost throughout the district in Sulawesi. Furthermore, leaf samples of *L. camara* Linn was cleaned and then dried in the open air. The dried samples were cut into small pieces and pulverized using a blender.

### 3.2 Extraction of secondary metabolites

A total of 2.5 kg of *L.camara* Linn leaf powdered were macerated for  $1 \times 24$  h as much as five times consecutively with chloroform. The extract was concentrated by evaporation so obtained viscous extract. Then put into dark bottles and stored in the refrigerator for analysis at later stage.

### 3.3 Fractionation with column chromatography

### Results

606





chloroform, and  
chloroform and aceton  
such as Liebermann  
Vagner, Dragendorff  
filter paper, alumina  
tissue, silica gel  
C plates of silica gel

will include sampl  
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Fractionation stage begins with the vacuum column chromatographic a technique (VCC) with a suitable solvent ratio is n-hexane and ethyl acetate (8:2). 13.48 grams chloroform extract was taken for VCC. Furthermore, the fractions of VCC were analyzed using thin layer chromatography (TLC) and combined based on similarity and stain patterns on the chromatogram.

## 4. Result and Discussion

### 4.1 Sample preparation and extraction

In testing the antibacterial activity of *L.camara* Linn leaf showed that chloroform extract has a high zone of inhibition against both gram positive and negative bacteria, namely *S.aureus* and *E.coli* (Dini *et al.*, 2010) and also refers to that used by the South Sulawesi community for the treatment of wounds is the leaf of the *L.camara* Linn plant. Hence in this study were selected as a sample of the leaf.

Maceration process to extract secondary metabolites compounds contained in the sample. Maceration carried out for 1 × 24 hours as much as five times that aims to maximize the maceration process to extract natural ingredient compounds contained in the sample. Furthermore, the extract was

concentrated with an evaporator to evaporate the solvent to obtain a chloroform extract of thick greenish-black with weighs 58.53 grams.

### 4.2 Fractionation of extract and antibacterial activity comparison

Fractionation is done to separate the compounds contained in the sample. Chloroform extract obtained was fractionated by VCC. Prior to the VCC, first extract was analyzed by TLC to determine a suitable eluent that can be used when making VCC. Based on the TLC results were done with several comparison eluent is obtained that shows the chromatogram separation stain or pattern that is both eluent n-hexane:ethyl acetate at a ratio of 8:2.

Fractionation performed with eluent with enhanced polarity starts from 100% n-hexane, n-hexane: ethyl acetate at a ratio of 19:1 to 1:19, ethyl acetate 100%, up to 100% methanol. The results obtained by VCC as many as 23 fractions.

Based on further analysis by TLC of the similarity obtained stain patterns on the chromatogram. VCC fractions results are merged to obtain six fractions combined as in Table 1.

Table 1. Combined Fraction of Fractionation VCC Results

Fraction	Combined Fraction	Weight (gr)
2-6	A	3.50
7-9	B	5.70
10-14	C	5.40
15-16	D	4.40
17-20	E	4.80
21-23	F	5.09

Results of previous studies indicate that the drug patent healed faster 3-4 days than chloroform extract of *L.camara* Linn (Muharram *et al.*, 2010). Therefore, it will

be screening the active compounds in the extract to eliminate inactive compound

Dini I *et al.* (2010) have carried out antibacterial test on fraction B and C by





examining the antibacterial activity against *S. aureus* and *E. coli* bacteria. The test results of the combined fractions are diameter of inhibition zone for each bacterium of *S. aureus* and *E. coli* in fraction B (18.3 mm and 12.4 mm) and fraction C (16.7 mm and 12.1 mm). And also the test results with the same bacteria on *L. camara* Linn leaf chloroform extract showed inhibition zone for bacteria *S. aureus* and *E. coli* in a row is 17.2 mm and 31.7 mm (Muharram *et al.*, 2010).

Comparison of the antibacterial activity results prove that there are certain compounds in the chloroform extract can inhibit the bacteria *S. aureus* and screening should be carried out first on all the fractions in order to obtain the active compounds to inhibit the bacteria *E. coli*.

#### 4.3 Screening and Identification of Compounds

Screening of the B combined fraction through three stages of fractionation, purification, and identification. Fractionations were performed using press column chromatography techniques. But prior to first find a suitable solvent for use in the column using TLC technique. The results of several replications and variations of solvent ratio is obtained that a good solvent to be used is n-hexane:ethyl acetate (9:1). Fractionation of B fraction (2.60 g) by press column chromatography produces 119 fractions. Then followed by TLC analysis with solvent ratio of n-hexane:ethyl acetate (9:1) and the combined fractions obtained 8 fractions are B1 fraction (fractions 1-9); B2 (fractions 10-14); B3 (fractions 15-22); B4 (fractions 23-30); B5 (fractions 31-56); B6 (fractions 57-80); B7 (fractions 81-108); and B8 (fractions 109-119).

Fractionation results showed only B4 and B8 fraction that form of crystals after the solvent was evaporated. Hence,

fractions of B4 and B8 respectively performed recrystallization with chloroform and ethyl acetate, and then analyzed by TLC with hexane:ethyl acetate eluent (9:1) and obtained a single stain. Compounds in B4 fraction as compound of 1 in the form of needle-shaped crystals and B8 fraction as compound of 2 is irregular crystalline or greenish white flakes.

Identification of B4 and B8 fraction aimed to determine the characteristics of the obtained compound. Identification of both is done by including:

- Specific reagents test of secondary metabolites group using reagent of  $\text{FeCl}_3$ , Liebermann-Burchard, Wagner, and Dragendorff. Compound of 1 positive on Liebermann-Burchard reagent with the formation of solution color changes to orange. Compound of 2 showed a positive reaction to the reagent  $\text{FeCl}_3$  with the formation color change of the test solution to yellow brown.
- Spectroscopic measurements of compounds intended as the data in structure elucidation. Spectroscopic measurements that have been carried out for compounds of 1 and 2 are FTIR spectroscopy. Figures 1 and 2 show the FTIR spectra for each of the compounds.

Compound of 1 is the class of triterpenoids, which contained carbonyl groups are shown in IR absorption at  $1690.63 \text{ cm}^{-1}$ , aliphatic C-H at  $2947 \text{ cm}^{-1}$ .

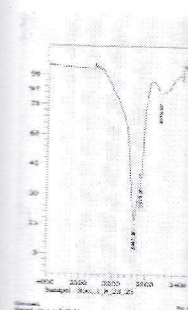


Figure 1. FTIR

The FTIR compound of 1 from the respective wave numbers follows:

Table 2. Compound 1

No.	Wavenumber
1.	2947
2.	1690
3.	1459
4.	1382

Figure 2 chromatogram compound of 2 as absorption peaks identified as follows:

Table 3. Compound 2

No.	Wavenumber
1.	3278
2.	2947
3.	1706
4.	1467



and B8 respectively. The crystallization of ethyl acetate, and the TLC with eluent (9: 1) to stain. Compounds in the form of 1 in the form of crystals and B8 fractions of irregular crystalline.

of B4 and B8 fractions are the characteristics of compound 1. Identification of compound 1 including:

tests of secondary group using reagent of Liebermann-Burchard, Wagner-Dorff. Compound of 1

formation of solution changes to the color of 2 showed a reaction to the reagent. Formation color change from yellow to brown.

measurements of compound 1 is the class of compounds that contain carbonyl group. IR absorption at 1706.05 cm<sup>-1</sup> and aromatic C-H at 2947 cm<sup>-1</sup>.

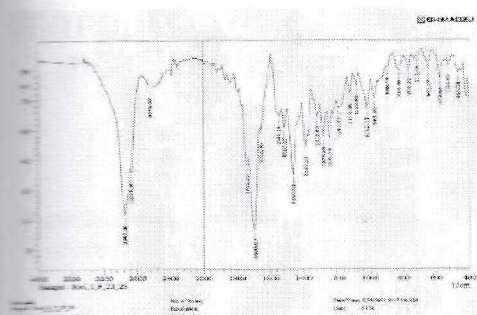


Figure 1. FTIR spectra of compound of 1

The FTIR chromatograms shows compound of 1 functional group as seen from the respective absorption peaks at wave numbers that are identified as follows:

Table 2. Compound of 1 functional groups interpretation

No.	Wavenumber (cm <sup>-1</sup> )	Functional groups
1.	2947.36	C-H
2.	1690.63	C=O
3.	1459.33	-CH <sub>2</sub> -
4.	1382.23	-CH <sub>3</sub>

Figure 2 shows the FTIR chromatogram of functional groups compound of 2 as seen from the respective absorption peaks at wave numbers that are identified as follows:

Table 3. Compound of 2 functional groups interpretation

No.	Wave number (cm <sup>-1</sup> )	Functional groups
1.	3278.89	-OH
2.	2947.36	-CH
3.	1706.05	C=O
4.	1467.04	-CH <sub>2</sub> -

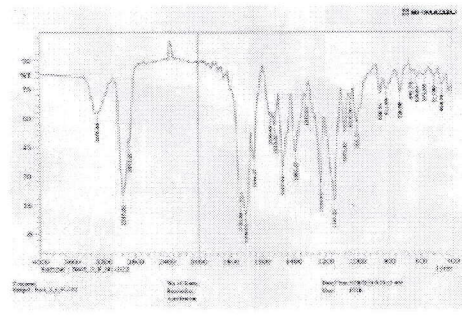


Figure 2. FTIR spectra of compound of 2

The compound of 2 was flavonoid compounds containing hydroxyl group at wave number of 3278.89 cm<sup>-1</sup>, carbonyl at 1706.05 cm<sup>-1</sup>, and aromatic at 1513.30 cm<sup>-1</sup> wave number.

The FTIR spectroscopy measurements have not been sufficient for the elucidation of the structure, so it will be another spectroscopic measurements to determine the structure of compounds. And also measuring the bioactivity of the combined fractions that would otherwise be performed for screening secondary metabolites in order to obtain the active compounds in preventing infection in skin wounds.

## 5. Conclusion

The results of screening by fractionation and purification techniques discovered class of terpenoids and flavonoids compounds through the identification of FTIR spectroscopy. However, the need to test the bioactivity for comparison with the chloroform extract of the leaf of *L.camara* Linn.

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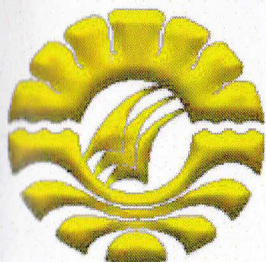
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### 1. Introduction

Copper is con  
most toxic metal an  
threat to the hu  
environment, even at l  
has been well r  
accumulation of copp  
causes brain, skin, l  
diseases. The permissi  
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# CERTIFICATE



Awarded to

Pince Salempa  
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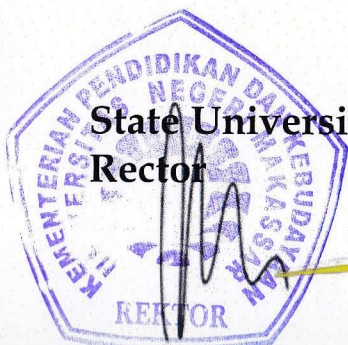
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FRACTION OF ROOT WOOD OF *PTEROSPERMUM SUBPELTATUM* C.B. ROB)

in

INTERNATIONAL CONFERENCE

ON MATHEMATICS, SCIENCES, TECHNOLOGY, EDUCATION , AND THEIR APPLICATIONS

*held on August 21, 2014, in the State University of Makassar,  
Makassar, South Sulawesi, Indonesia*



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